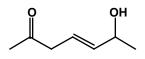
JASPERSE CHEM 360 TEST 3 VERSION 2 Ch 18 Ketones and Aldehydes Ch 22 Additions and Conensations of Enols and Enolate Ions

1. Nomenclature. Provide the structure or the name for the following. If stereochemistry is a factor, do not neglect it. (6 pt)

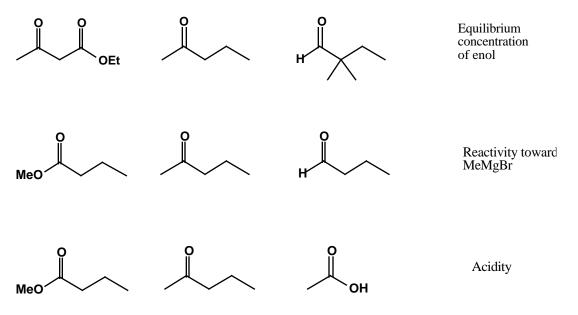


3-isopropylbenzaldehyde

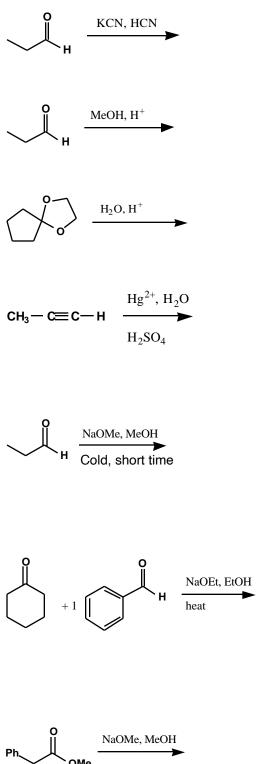
Ph H

optically active

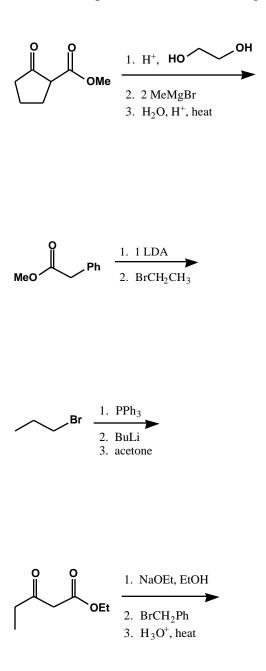
2. Rank the following, with 1 being highest, or most. (6 pt)



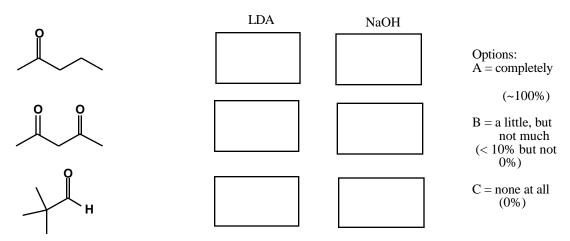
3. Draw the products for the following reactions (3 pt each)



4. Draw the products for the following multistep reactions. (3 pt each)

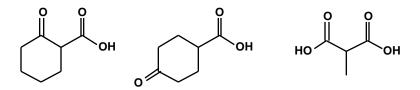


5. For the following chemicals, describe the extent to which each would be deprotonated by LDA $(LiN-iPr_2)$ or by NaOH at equilibrium. Fill in all 6 boxes. Options are complete deprotonation (A), a little deprotonation (B), and no deprotonation (C). (6 pt)

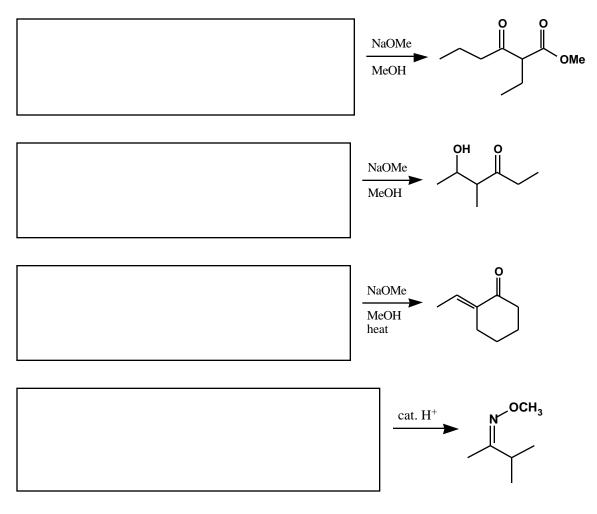


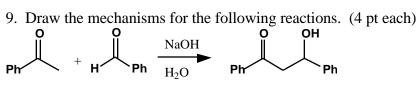
- 6. Suggest a plausible structure consistent with the following information. (5 pt)
- a. It reacts positively with 2,4-dinitrophenylhydrazine.
- b. It reacts positively with NaOH/I₂, the iodoform test
- c. It does not react with Tollen's reagent $[Ag(NH_3)_2^+OH^-]$.
- d. It does not react with Br_2 in dichloromethane solvent.
- e. Chemical formula is C7H12O
- f. It's ¹³C spectrum shows 5 carbons (1 singlet, 1 doublet, 2 triplets, and 1 quartet)

7. Which of the following would not undergo decarboxylation (loss of CO₂) upon heating? (2 pt)

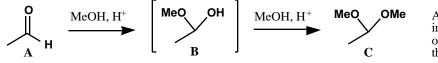


8. Put in the starting materials from which the following would be made. (3 each)

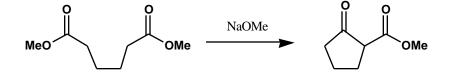


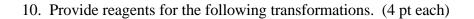


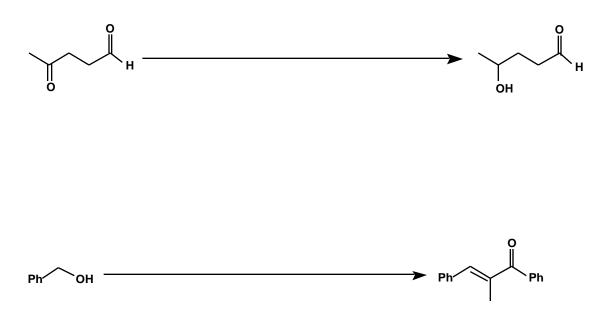
(Note: this one counts as 2 problems, 8 points total)



All steps are actually in equilibrium, but I only want you to show the forward direction







11. Design a synthesis for the following alkene **FROM ALCOHOLS WITH NO MORE THAN 5 CARBONS**. (6 pt)

